



Section A

Answers

- | | | | | | | |
|------|-------|-------|-------|-------|-------|-------|
| 1. C | 6. D | 11. D | 16. C | 21. C | 26. C | 31. C |
| 2. C | 7. A | 12. A | 17. B | 22. B | 27. B | 32. C |
| 3. B | 8. D | 13. D | 18. D | 23. B | 28. D | 33. A |
| 4. B | 9. D | 14. C | 19. C | 24. C | 29. A | |
| 5. C | 10. A | 15. C | 20. C | 25. C | 30. D | |

Solution

1. C

$$\theta = \frac{5.8 - 8.4}{3.6 - 8.4} \times 100 = 54^{\circ}\text{C}$$

2. C

- ✓ (1) Silver is a poor emitter of heat, but there is still some heat transfers through the vacuum by radiation.
- ✗ (2) As there is no air inside the double glass shell, heat cannot be transferred by convection.
- ✓ (3) Although the stopper is a poor conductor, some heat still transfers through the stopper by conduction.

3. B

Heat lost by the tea + heat lost by the cup = heat gained by the milk

$$m_t c_t \Delta T_t + C_c \Delta T_c = m_m c_m \Delta T_m$$

$$(0.320)(4200)(90 - \theta) + (250)(90 - \theta) = (0.060)(3800)(\theta - 15)$$

$$\therefore \theta = 80.6 \approx 81^{\circ}\text{C}$$

4. B

- ✓ (1) As the specific heat capacity of water is higher, the temperature rise is smaller.
- ✗ (2) Hot air above the beach rises up, cool air above the sea replaces, thus the direction of wind should blow from the sea to the beach.
- ✓ (3) Water evaporates from the sea and absorbs latent heat of vaporization from the water, this gives the cooling effect to the water.

5. C

- ✗ (1) As the volume of the container is fixed, the average separation is unchanged.
- ✓ (2) As the temperature increases, the speed of the molecules increases, thus the change of momentum increases in hitting the walls.
- ✓ (3) As the speed increases, the gas molecules hit the walls more frequently.

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