

Name: **KEY** Date: _____

Class/Period: _____

Honors Chemistry
Final Exam Review Sheet Fall 2013



The fall semester final exam is 15% of your overall final grade. The multiple-choice exam will consist of 100 multiple choice questions answered on a scantron and will be given on _____. To study for the multiple choice portion, it is suggested you use this review sheet, your notes, handouts, textbook, labs, and your Performance Final review (the same topics are covered). On the day of the exam, you will need to bring a #2 pencil and a calculator. A periodic table will be provided for you.

Scientific Method and Measurement

- What is a hypothesis? **Testable prediction about something that has been observed**
 - What is a theory? **Explanation supported by many experiments**
 - What is a law? **Summary of accepted facts about nature (law of gravity)**
- What is the difference between accuracy and precision? **Accuracy refers to how close a measured value is to an actual value. Precision refers to how close a series of measurements are to each other.**
 - Determine the number of significant figures in the following numbers to show precision:

1) 5370 3	3) 240.10 5	5) 431.801 6
2) 0.00369 3	4) 2001 4	6) 10 235.0 6
- Use the data table below to discern whether the students' measurements are *accurate only*, *precise only*, *both accurate and precise*, or *neither accurate nor precise*.

	Measurement 1	Measurement 2	Measurement 3	Measurement 4
Jameel	5.0 g	4.9 g	5.0 g	5.1 g
Cody	8.0 g	8.2 g	8.0 g	7.9 g
Rachel	2.0 g	12.0 g	8.0 g	20.0 g
Juan	10.0 g	9.9 g	10.1 g	9.9 g

*****The accepted value of the rock's mass is 5.0 g

- Describe Jameel's measurements. **Accurate and precise**
 - Describe Cody's measurements. **Precise, but not accurate (Precise only)**
 - Describe Rachel's measurements. **Neither accurate or precise**
 - Describe Juan's measurements. **Precise, but not accurate (Precise only)**
- What piece of lab equipment measures mass? **balance**
 - What piece of lab equipment most accurately measures volume of a *liquid*? **Graduated cylinder**
 - What is the formula for density? **$D = \frac{\text{mass}}{\text{volume}}$**
 - How is the volume of a regularly-shaped object, such as a box or book, found?
Length (cm) x width (cm) x height (cm) = volume (cm³)



2013 Chemistry Final Exam Review Sheet

ML Morrison



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