

Worksheet 1.6

Supplemental Instruction
Iowa State University

Leader: Lea

Course: Chem 177

Instructor: Dr. Schewe-Miller

Date: 2/5/2013

- 1.) A piece of aluminum foil 1.00 cm square and 0.550 mm thick is allowed to react with bromine to form aluminum bromide.

a.) How many moles of aluminum were used? (density Al = 2.699 g/cm³)

$$.550 \text{ mm} \times \frac{1 \text{ cm}}{10 \text{ mm}} = .055 \text{ cm}$$

$$1 \text{ cm} \times 1 \text{ cm} \times .055 \text{ cm} = .055 \text{ cm}^3 \times \frac{2.699 \text{ g}}{\text{cm}^3} = .148445 \text{ g Al} \times \frac{1 \text{ mol Al}}{26.98 \text{ g Al}} = .005502 \text{ mol Al}$$

$5.50 \times 10^{-3} \text{ mol Al}$

b.) How many grams of aluminum bromide form, assuming the aluminum reacts completely? What type of reaction is this?



$$5.50 \times 10^{-3} \text{ mol Al} \times \frac{2 \text{ mol AlBr}_3}{2 \text{ mol Al}} \times \frac{266.69 \text{ g AlBr}_3}{1 \text{ mol AlBr}_3} = 1.47 \text{ g}$$

composition reaction

2.) The chemical formula for benzene is C₆H₆.

a. Determine the percent composition of H and C for benzene.

$$6(12.01) + 6(1.01) = 78.12$$

$$\text{C}: \frac{72.06}{78.12} \times 100 = 92\% \quad \text{H}: 100 - 92\% = 8\%$$

$\text{C}: 92\% \\ \text{H}: 8\%$

b. Write a balanced combustion reaction for benzene.



c. Assuming you have 364g of benzene, how many moles of O₂ will be required to completely combust this amount of benzene?

$$364 \text{ g} \times \frac{1 \text{ mol C}_6\text{H}_6}{78.12 \text{ g C}_6\text{H}_6} \times \frac{15 \text{ mol O}_2}{2 \text{ mol C}_6\text{H}_6} = 35.0 \text{ mol}$$

d. How many grams of CO₂ will this produce? How many grams of H₂O?

$$\text{CO}_2: 364 \text{ g C}_6\text{H}_6 \times \frac{1 \text{ mol C}_6\text{H}_6}{78.12 \text{ g C}_6\text{H}_6} \times \frac{12 \text{ mol CO}_2}{2 \text{ mol C}_6\text{H}_6} \times \frac{44.01 \text{ g CO}_2}{1 \text{ mol CO}_2} = 1.23 \times 10^3 \text{ g CO}_2$$
$$\text{H}_2\text{O}: 364 \text{ g C}_6\text{H}_6 \times \frac{1 \text{ mol C}_6\text{H}_6}{78.12 \text{ g C}_6\text{H}_6} \times \frac{6 \text{ mol H}_2\text{O}}{2 \text{ mol C}_6\text{H}_6} \times \frac{18 \text{ g H}_2\text{O}}{1 \text{ mol H}_2\text{O}} = 2.52 \times 10^2 \text{ g H}_2\text{O}$$

e. You perform the combustion, and 956g of CO₂ are produced. What is the percent yield?

$$\frac{\text{Actual}}{\text{Theoretical}} \times 100 = \frac{956 \text{ g CO}_2}{1230 \text{ g CO}_2} \times 100 = 77.7\%$$

4 Practice Problems Chemistry Answers

David A. Ucko



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but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity

Survival Guide to Organic Chemistry Patrick E.

McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 Reviews key general chemistry concepts and techniques adapted for application to important organic principles Provides practical guidance to help students make the notoriously well known and arduous transition from general chemistry to organic chemistry Explains organic concepts and reaction mechanisms generally expanding the focus on how to understand each step from a more intuitive viewpoint Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms Implements a user friendly process to achieve the end result of problem solving Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum features practice problems and solutions to help master the material including an extensive and comprehensive bank of practice exams with solutions

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

Ebook: Introductory Chemistry: An Atoms First Approach Burdge, 2016-04-16

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for Chemistry 2nd Edition Disha Experts,2019-09-25 **Chemical Literacy and Writing Chemical Reactions** Nikolay Gerasimchuk,2022-09-12 Writing chemical reactions in general and inorganic chemistry is not a trivial task However writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject Unfortunately very few students grasp the concept of the correct writing of chemical reactions quickly and so are unable to move through topics of general analytical and inorganic chemistry freely Because the ability to write and balance different types of chemical reactions is a fundamental issue this becomes a key question of chemical literacy The successful writing of chemical reactions includes two components the prediction of products of these reactions and their possible variations and balancing these reactions providing a material balance between starting compounds and reactions products This book explores that element of the teaching of the fundamentals of chemical literacy writing complete equations of chemical reactions and balancing them It contains 49 figures 22 schemes and 12 tables and 93 problems with answers This book will be very useful for high school students interested in chemical sciences higher education teachers students in colleges and universities majoring in chemistry and biochemistry and chemistry professional working in industry It also contains information about properties of the most common elements and applications of a variety of their chemical compounds *Ebook: Chemistry* Julia Burdge,2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text Chemistry Workbook For Dummies Peter J. Mikulecky,Chris Hren,2014-11-26 Hundreds of practice problems to help you conquer chemistry Are you confounded by chemistry Subject by subject problem by problem Chemistry Workbook For Dummies lends a helping hand so you can make sense of this often intimidating subject Packed with hundreds of practice problems that cover the gamut of everything you ll encounter in your introductory chemistry course this hands on guide will have you working your way through basic chemistry in no time You can pick and choose the chapters and types of problems that challenge you the most or you can work from cover to cover With plenty of practice problems on everything from matter and molecules to moles and measurements Chemistry Workbook For Dummies has everything you need to score higher in chemistry Practice on hundreds of beginning to advanced chemistry problems Review key chemistry concepts Get complete answer explanations for all problems Focus on the exact topics of a typical introductory chemistry course If you re a chemistry student who gets lost halfway through a problem or worse yet doesn t know where to begin Chemistry Workbook For Dummies is packed with chemistry practice problems that will have you conquering chemistry in a flash Chemistry in the Community. American Chemical Society,2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts Basics for

Chemistry David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material **General Chemistry** Donald

A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value **AP Chemistry Premium, 2025: Prep Book with 6 Practice Tests +**

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