Chapter 1 - Introduction to Chemistry - Answer Key

Chapter 1: 1, 3, 5, 8, 9, 10, 12 - 19, 21, 22, 30, 36 (18 total)

Section Review 1.1

- 1. Match the numbered terms in the right column with the lettered terms in the left column.
- 1. Life
- Matter and its changes
- Carbon
- 4. Applications
- Composition
- 3. Explain the difference between pure and applied chemistry.

Pure chemistry examines chemical aspects of the world for the sake of knowledge alone; applied chemistry uses this knowledge to solve real-world problems.

Section Review 1.2

- Name at least three areas of science in which chemistry plays an important role.
 - a, agriculture
 - b. medical and biotechnological sciences
 - c. materials science

Section Review 1.3

8. Describe the steps involved in the scientific method. Provide an example.

Observation: My stamach is growling

Hypothesis: Maybe I'm hungry

Experiment: Eat something and see what happens

9. Distinguish between a theory, a hypothesis, and a scientific law.

A law is a concise statement that summarizes the results of a broad variety of observations and experiments. A hypothesis is a proposed explanation or reason for what is observed. A theory is an extensively tested explanation for experimental results.

10. Why should a hypothesis be developed before experiments take place?

The design of the experiments will be guided by the proposed hypothesis Veritas/ Chemistry/ Chapter 1 – Introduction to Chemistry

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