

19.5

HYDROGEN IONS AND ACIDITY

Section Review

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH and pOH and concentrations of acids and bases and vice versa.
3. Calculate the pH of weak acids.

Vocabulary

- | | | |
|-------------------|-----------------------------|-----------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. acid solution |
| 4. autoionization | 5. acid rain | 6. acid rain solution |
| | 7. acid solution | 8. pH |

Key Equation

- | | |
|------------------------------------|------------------------------------|
| 1. $\text{pH} = -\log[\text{H}^+]$ | 2. $\text{pH} = -\log[\text{H}^+]$ |
| 3. $\text{pH} = -\log[\text{H}^+]$ | 4. $\text{pH} = -\log[\text{H}^+]$ |

Part A Completion

Write the equation and/or the numerical value(s) of the quantity indicated that are indicated in the problem. Be sure to include units in your answers.

1. What substance is _____, a strong hydrogen ion donor?
2. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
3. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
4. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
5. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
6. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
7. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
8. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
9. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
10. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
11. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
12. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
13. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
14. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
15. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
16. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
17. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
18. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
19. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.
20. _____ is the concentration of H^+ ions in pure water at 25°C. _____ is _____.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

11. _____

12. _____

13. _____

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17. _____

18. _____

19. _____

20. _____

19 2 Hydrogen Ions And Acidity Section Review Answers

Roman Wölfel



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