Multiple Choice: (from Zumdahl Chemistry, 9th Edition, AP Edition)

Refer to the following diagram for questions 1-3

A 100.0 mL Graduated Cylinder is filled with 100.0 mL of water. A few drops of food colouring are added to the top of the water, without mixing. Initially, the top of the graduated cylinder appears to be very dark, while the bottom remains clear and colourless. After 5 min, the liquid in the graduated cylinder appears to be uniformly coloured.

- Classify the contents in the graduated cylinder before any food colouring is added.
- (A) Element (B) Compound. (C) Heterogeneous mixture
- (D) Homogeneous mixture
- Classify the contents in the graduated cylinder just after the food colouring is added.
- (A) Element (B) Compound. (C) Heterogeneous Mixture (D) Homogeneous mixture
- 3. Classify the contents in the graduated cylinder after the 5 minutes has elapsed.
- (A) Element (B) Compound. (C) Heterogeneous Mixture (D) Homogeneous mixture
- 4. A 10.00 g piece of metal is submerged in a graduated cylinder initially containing 20.00 mL of water. With the metal in the cylinder, the level of water is recorded as 21.40 mL. Determine the identity of the metal.
- (A) Al ($\rho = 2.70 \text{ g/mL}$)
- (B) $Zn (\rho = 7.13 \text{ g/mL})$
- (C) Ag ($\rho = 10.49 \text{ g/mL}$)
- (D) Au ($\rho = 19.32 \text{ g/mL}$)
- 239 naPu → X + a

In the decay represented above, what is the nucleon number of particle X?

- (A) 92 (B) 96 (C) 235 (D) 243
- 6. Which of the following statements about atoms is CORRECT?
 - (A) Atoms are charged because they have different numbers of protons and neutrons>
 - (B) All atoms of an element have the same number of nucleons and electrons.
 - (C) An atom is mostly filled with protons.
 - (D) The nucleus is negatively charged.
- When each of the following is heated to 50°C and the temperature is held constant for 5 min, which one undergoes only a physical change.
 - (A) egg (B) steak (C) ice cream (D) cake batter
- 7. A sample of Ti was ionized by electron impact in a mass spectrometer. Information from the mass spectrum about the isotopes of Ti in the sample is shown in Table 2.

Table 1. Results from mass spectrum of Ti sample				
me/z	46	47	48	49
Percent Abundance	9.1	7.8	74.6	8.5

BIEFORE.

(DATAMENTO)

producted as finder

APTER

Calculate the relative atomic mass of Ti in this sample. Report answers to one decimal place.

Multiple Choice Review Zumdahl Ninth Edition

Peipei Pang



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texture

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conventional method for designing surface finish as per is 696 surface texture specified by indicating the following a roughness value i e ra value in mm e machining allowance in mm e sampling length or instrument cut off length in mm e machining production method and e direction of lay in the symbol form as e m e r

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for jis symbols specify roughness ra roughness rz rmax for gost symbols you can also select use for notation displays the surface finish symbol 0 5 times larger than the default size add default symbol displays the default surface finish symbol in parentheses format use document font

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bs en iso 1302 2002 geometrical product specifications gps indication of surface texture in technical product documentation for notes on surface roughness use link surface roughness notes the symbol indicates the surface finish requirements and shows a machining allowance requirement of 3mm on all surfaces

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surface symbol symbol indicating a surface where removal processes are prohibited symbol indicating a surface that requires a removal process examples of indicating the ra upper limit a b c example of indicating grain direction examples of indicating ra upper limit and lower limit a b examples of indicating the machining method

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