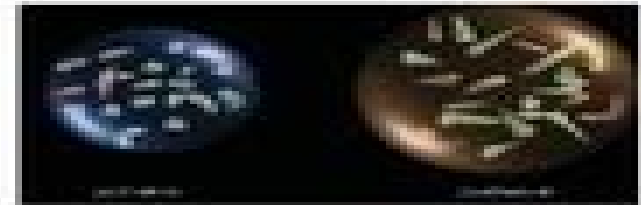


Lesson 1: Temperature and Heat Transfer

Temperature and Heat Transfer

- Temperature—the quantity that tells us how _____ or _____ something is. OR a measure of the average _____ of the atoms and molecules
- Kinetic energy—the energy of _____
- As temperature increases, pressure increases; as temperature decreases, pressure decreases
 - Warm, less dense air OR cold, more dense air
 - Absolute zero (-273°C or -459°F) atoms and molecules have no energy
- _____—energy in the process of being transferred from one object to another because of the temperature difference between them



Temperature Scales

- Kelvin Scale—absolute scale with absolute 0; no negative numbers
- Fahrenheit Scale—water freezes at 32; water boils at 212
- Celsius Scale—water freezes at 0; water boils at 100
- Conversions (do not need to memorize)
 - $^{\circ}\text{C} = 5/9 (^{\circ}\text{F} - 32)$
 - $\text{K} = ^{\circ}\text{C} + 273$

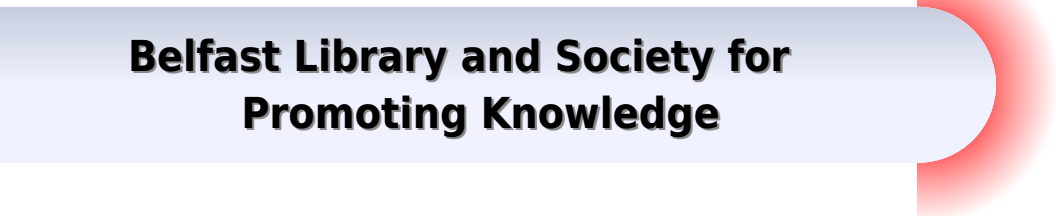
Latent Heat—The Hidden Warmth

- Water vapor is invisible until it turns into _____ or _____ through a **phase change or change of state**.
- _____—the heat energy required to change a substance from one state to another
- Evaporation is a _____ process.
 - In a drop of water, molecules are constantly escaping. As the outside molecules escape, the average motion of the remaining particles _____—lower temperature
 - Example: Body heat can cause the energy for _____ to evaporate which then cools the body.
 - Energy is _____ or hidden in the water vapor making it latent heat.
- _____ is a warming process
 - Example: during condensation, the energy released from water vapor is _____ by a cold pop can



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