

2nd Semester CHEMISTRY Final Exam Review Guide 2014-2015

What will the test look like?

- 100 multiple choice problems
- All 2nd semester units, all material will be covered (see list below)
- 2 hour time period, no extra time. Some will finish early, some will use whole 2 hours.
- Will go in test category, will be worth roughly 10% of overall grade

The units we have covered are listed below with the **approximate** percentage of questions on the final exam that will be from that unit:

Chapter	Concept(s)	Percentage of Q's
6 & 7	Ionic, Metallic, and Covalent Bonding Formulas, Shapes, Polarity, Intermolecular Forces	25
End of 7	Empirical & Molecular Formulas Percent Composition	5
8 & 9	Chemical Reactions Stoichiometry	30-40
10 & 11	Gas Laws Stoichiometry with Gases	25-30

- You will get all the charts, tables, and gas law equations and constants we have used in class so far: solubility table, activity series, electronegativity etc...
- What you were required to memorize, you will still be required to have memorized: polyatomic ions, common elements, etc....
- As with other tests, you will be allowed to use a scientific non-graphing calculator. There are a limited number of these you may borrow (not enough for everyone). Be prepared and bring one with you to class. You will also need #2 pencils for the scantron.
- You will not get extra time beyond the two hour time block reserved for this final. If you should finish your final early, please bring something with you to keep you occupied (a magazine, etc.)
- As usual, you may not have cell phones, ipods, headphones, etc.... out during the test.
- You may not leave the room while you are taking the test. Therefore, go to the bathroom **BEFORE** the test starts. If you think you will get thirsty, bring a water bottle with you.
- If you know you will be absent, schedule a make-up time with me. Please make these arrangements as soon as possible.

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Clifford Lane



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