# Brush border membranes



# **Brush Border Membranes Symposium**

**Roman Wölfel** 

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web mar 13 2023 the measured values  $\tau$  ign is represented by the data for n decane t 1600 k at p 3 5 atm n decane t 1175 k p 5 atm n decane t 1300 k p 13 atm n decane and kerosene jet a t 1750 k p 9 atm jet a kerosene t 1320 k p 11 atm and jet a kerosene t 1380 k p 10 atm

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web allow thegaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of kerosene nevertheless in actual calculations involving liquid kerosene fuel the heat capacity and latentheat of vaporization of liquid kerosene 12 are used for

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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web compiled by g izquierdo rodríguez page 2 1 directly proportional to the product of their charges f q 1 q 2 2 inversely proportional to square of the distance between them 2 1 r f therefore f  $\alpha$  r2 kqq coulomb s law in symbols 2 1 2 r kq q f coulomb s law the magnitude of the electrostatic force exerted by one point charge q

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