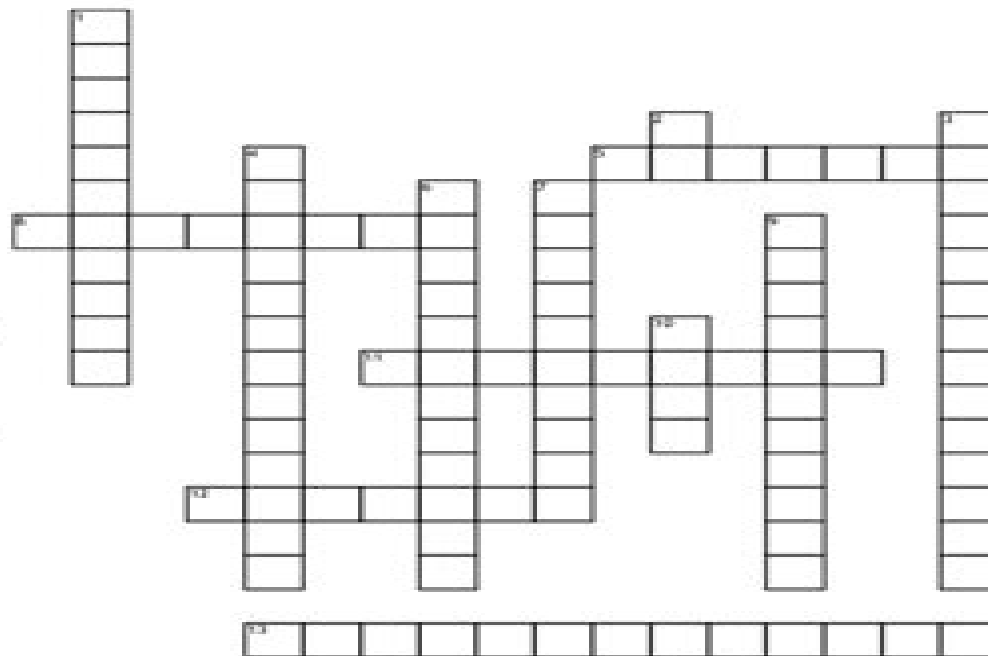


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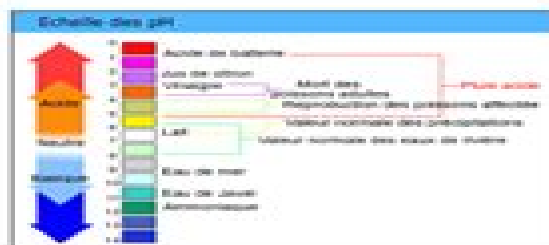
Across

5. a logarithmic scale that measures the acidity or alkalinity of a solution on a scale of 1 to 14
8. the concentration of a substance in moles per liter of solution
11. a substance that undergoes a color change that can be used to determine when a reaction is complete
12. a measure of the hydrogen (hydronium) ion concentration of a solution
13. a substance that produces hydroxide ions (OH^-) as the only negative ions when dissolved in water



Down

1. a substance whose water solution conducts an electric current
2. a logarithm of a solution's hydrogen ion concentration
3. the reaction between an acid and a base to produce water and salt
4. a substance that produces hydronium ions (H_3O^+) as the only positive ions when dissolved in water
6. A compound formed by the combination of water with a hydrogen ion, H_3O^+
7. a measure of the hydroxide ion concentration of a solution
9. a hydrogen atom without it's electron (consisting solely on a proton)
10. the product other than water of a neutralization reaction; an ionic substance consisting of a metallic cation and anion other than the hydroxide ion



Acids And Bases Crossword Answer Key

Judd E. Hollander



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Acids and Bases Ronald Percy Bell,1952

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