

59. Alpha particles with their high ionizing charge but low penetrating ability can best be shielded by?

a. paper

b. clothing

c. water

d. lead

Alpha α particles - paper is a sufficient shield

Beta β particles - heavy clothing is sufficient to shield

Gamma rays - lead is required to shield

↔ = polar

↕ = polar

↗ = non-polar

↔ = non-polar

↕ = non-polar

60. The half-life of Se-121 is 10 days. If you started with 40 grams of this isotope, how much would you have left after 30 days?

a. 10 g

b. none

c. 15 g

d. 5 g

Mass of Se-121	Number of half-lives	Time
40 g	0	0
20 g	1	10 days
10 g	2	20 days
5 g	3	30 days

⬆ = particles go faster

⬇ = particles go slower

PERIODIC TRENDS

ATOMIC RADIUS: decreases →
increases ↓

METALLIC CHARACTER:

decreases ↓
increases →

IONIZATION ⬆ decreases ↓
increases →

A → mass # (# protons + # neutrons)
Z → atomic # (# of protons)
X → chemical symbol

• Isotopes = atoms with same # of protons, but different # of neutrons

EXAMPLE: Ne-20 → $^{20}_{10}\text{Ne}$
Ne-21 → $^{21}_{10}\text{Ne}$
Ne-22 → $^{22}_{10}\text{Ne}$

• Dilution Equation $M_1 V_1 = M_2 V_2$

• $\text{mass\%} = \frac{\text{mass solute}}{\text{mass of solution (solute + solvent)}} \times 100\%$

• $\text{Molarity (M)} = \frac{\text{mol solute}}{\text{L solution}}$

• $\text{molality (m)} = \frac{\text{mol solute}}{\text{kg solvent}}$

• #protons - #electrons = Charge #

• #neutrons = A (mass #) - Z (atomic #)

• if no charge, the #protons + #electrons = same

• if + (positive) charge (cation) = #electrons is less than #protons

• if - (negative) charge (anion) = #electrons is greater than #protons

• Protons ≠ Charge → they give identity to the element.

Acs General Chemistry Exam Review

Douglas R. Whitney, Andrew G. Malizio



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contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

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[Acing the ACS](#) James Coy-Dibley, 2017-09-05 Study guide for first year general chemistry ACS Exam

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