

Maximum Marks: 80

Time allowed Two hours

Answers to this Paper must be written on the paper provided separately

You will not be allowed to write during first 15 minutes.

This time is to be spent in reading the question paper.

The time given at the head of this Paper is the time allowed for writing the answers.

Section A is compulsory. Attempt any four questions from Section B.

The intended marks for questions or parts of questions are given in brackets [].

SECTION A (40 Marks)

(Attempt all questions from this Section.)

Question 1

Choose the correct answers to the questions from the given options.

[15]

(Do not copy the questions, write the correct answers only.)

(i) An element in period 3, whose electron affinity is zero:

- (a) Neon
- (b) Sulphur
- (c) Sodium
- (d) Argon

Answer: (d)

Argon belongs to period 3. It is an inert element and stable. It neither requires gain nor loss. It has its octet complete and hence it has the zero electron affinity.

(ii) An element with the largest atomic radius among the following is:

- (a) Carbon
- (b) Nitrogen

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Table of Contents 13 Grade 10 Chemistry Final Exam Paper And Memorandum

1. Understanding the eBook 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - The Rise of Digital Reading 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Advantages of eBooks Over Traditional Books
2. Identifying 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - User-Friendly Interface
4. Exploring eBook Recommendations from 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Personalized Recommendations

- 13 Grade 10 Chemistry Final Exam Paper And Memorandum User Reviews and Ratings
- 13 Grade 10 Chemistry Final Exam Paper And Memorandum and Bestseller Lists
- 5. Accessing 13 Grade 10 Chemistry Final Exam Paper And Memorandum Free and Paid eBooks
 - 13 Grade 10 Chemistry Final Exam Paper And Memorandum Public Domain eBooks
 - 13 Grade 10 Chemistry Final Exam Paper And Memorandum eBook Subscription Services
 - 13 Grade 10 Chemistry Final Exam Paper And Memorandum Budget-Friendly Options
- 6. Navigating 13 Grade 10 Chemistry Final Exam Paper And Memorandum eBook Formats
 - ePub, PDF, MOBI, and More
 - 13 Grade 10 Chemistry Final Exam Paper And Memorandum Compatibility with Devices
 - 13 Grade 10 Chemistry Final Exam Paper And Memorandum Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Highlighting and Note-Taking 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Interactive Elements 13 Grade 10 Chemistry Final Exam Paper And Memorandum
- 8. Staying Engaged with 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 13 Grade 10 Chemistry Final Exam Paper And Memorandum
- 9. Balancing eBooks and Physical Books 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 13 Grade 10 Chemistry Final Exam Paper And Memorandum
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Setting Reading Goals 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 13 Grade 10 Chemistry Final Exam Paper And Memorandum
 - Fact-Checking eBook Content of 13 Grade 10 Chemistry Final Exam Paper And Memorandum

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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