

CAMBRIDGE INTERNATIONAL EXAMINATIONS
GCE Advanced Level

MARK SCHEME for the May/June 2013 series

9706 ACCOUNTING

9706/43

Paper 4 (Problem Solving – Supplement),
maximum raw mark 120

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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a

conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3

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an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of

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acids bases and enzymes many acids and bases in living things provide the pH that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper pH for enzymes to do their work

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acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

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a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

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