

Section 9.2: Team Learning Worksheet

1. Consider the following unbalanced equation:



You have 5.00 grams of manganese(IV) oxide:

- How many moles of manganese(IV) oxide do you have?
- How many moles of HCl do you need for a complete reaction?
- How many grams of HCl do you need for a complete reaction?
- How many grams of each product (H_2O , MnCl_2 , and Cl_2) would be formed?
- Prove that mass has been conserved in this reaction.
- Is the number of atoms conserved? Support your answer.
- Is the number of molecules conserved? Support your answer.

20 2 Team Learning Worksheet Answer

**S Joubert, P Laporte, A McFarland, M
Oosten, H Vrooman**



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